

# LEPTORHABINE - A NEW ALKALOID

FROM *Leptorhabdos parviflora*

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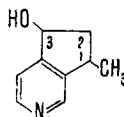
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We have investigated the epigeal part of the plant *L. parviflora* collected in the valley of the R. Talas (close to the town of Dzhambul) in the flowering period. When the combined ether-soluble alkaloids were separated on a column of alumina, from ether-chloroform (10:2) eluates we isolated a liquid base with the composition  $C_9H_{11}NO$ ,  $[\alpha]_D^{20} +10^\circ$  (c 1.015; chloroform);  $M^+$  149;  $R_f$  0.56 [TLC on silica gel in the benzene-methanol (4:1) system]. The UV spectrum of the base has two maxima at  $\lambda_{max}$  263, 269 nm ( $\log \epsilon$  3.36, 3.33) which are characteristic for alkaloids of the pyridine type [1-4]. In the IR spectrum of the base, absorption bands appear at  $3400-3200\text{ cm}^{-1}$  (OH),  $2980-2940\text{ cm}^{-1}$  (C-CH<sub>3</sub>), and  $1605$  and  $1580\text{ cm}^{-1}$  (pyridine ring). The base is a new one and we have called it leptorhabine.

In the NMR spectrum of leptorhabine (JNM-4H-100/100 MHz in CCl<sub>4</sub>, internal standard HMDS,  $\delta$  scale) a one-proton singlet can be clearly seen at 8.11 ppm and also two one-proton doublets at 8.07 and 7.15 ppm ( $J=5.0$  Hz) corresponding to three aromatic hydrogen atoms in the  $\alpha$ -,  $\alpha'$ -, and  $\beta'$ -positions with respect to the nitrogen atom of the pyridine ring. The absence of other signals in this region and the production of pyridine-3,4-dicarboxylic acid by the oxidation of the base with KMnO<sub>4</sub> in an alkaline medium shows the positions of the substituent in the pyridine ring.

The one-proton signal at 6.94 ppm from the proton of the hydroxy group is shifted downfield, which is explained by the presence of a hydroxy group on a carbon atom directly conjugated with an aromatic nucleus. A three-proton doublet at 1.20 ppm ( $J=7.0$  Hz) is due to the protons of a methyl group, and a two-proton multiplet at 1.97 ppm to two methylene protons. Two one-proton multiplets at 5.06 and 3.30 ppm relate to the two methine protons at C<sub>1</sub> and C<sub>3</sub>.

In the NMR spectrum of O-acetylleptorhabine the signal of the proton of the hydroxy group has disappeared and a three-proton singlet has appeared at 1.92 ppm from the CH<sub>3</sub>-C=O group, and the signal of one methine proton has shifted from 5.06 ppm to 5.99 ppm.



The mass spectrum of leptorhabine shows peaks of ions with  $m/e$  149 ( $M^+$ ), 132, 131, 118, 117, 106, 104, 79, 77, 65, 63. What has been said above has enabled us to suggest for leptorhabine two possible structures: 1-hydroxy-3-methylcyclopenta[d]pyridine and 1-methyl-3-hydroxycyclopenta[d]pyridine. From a comparison of the NMR spectrum of leptorhabine with literature data [3, 4], a consideration of Dreiding models [descreening influence of the acetyl group on the  $\alpha$ - and  $\alpha'$ -protons ( $\delta$  8.34; 8.25 ppm) and the absence of such screening for the  $\beta'$ -proton (7.17 ppm)] and also from biogenetic considerations, we consider the above structure to be most probable.

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